

Write correct formulas of the compounds formed when the positive ions in the vertical column combine with the negative ions listed across the top row. The first two are done for you.

	nitrate	sulfate	carbonate	phosphate	hydroxide	chromate
sodium	$\text{NaNO}_3$	$\text{Na}_2\text{SO}_4$				
silver						
ammonium						
mercury(I)						
zinc						
calcium						
magnesium						
copper(I)						
lead(II)						
aluminum						
manganese(III)						
cobalt(III)						
cupric						
ferric						
plumbous						
potassium						
barium						

	hydrogen carbonate	dichromate	acetate	sulfide	chloride	sulfite
sodium	$\text{NaHCO}_3$	$\text{Na}_2\text{Cr}_2\text{O}_7$				
silver						
ammonium						
mercury(II)						
tin(II)						
calcium						
magnesium						
copper(II)						
lead(II)						
aluminum						
manganese(III)						
cobalt(III)						
cuprous						
ferrous						
plumbic						
potassium						
barium						